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Hydrogen + oxygen → water $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ B. Write a balanced chemical equation for the reaction of carbon dioxide with limewater. The formula for the active substance in limewater is on the pipet, and the products of the reaction are calcium carbonate and water. $\text{Ca}(\text{OH})_2(\text{aq}) + \text{CO}_2(\text{g}) = \text{CaCO}_3(\text{s}) + \text{H}_2\text{O}(\text{l})$ Good job with the equations.

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